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LETTERS

Synthesis of triazolopyridines as conformationally constrained tertiary amides

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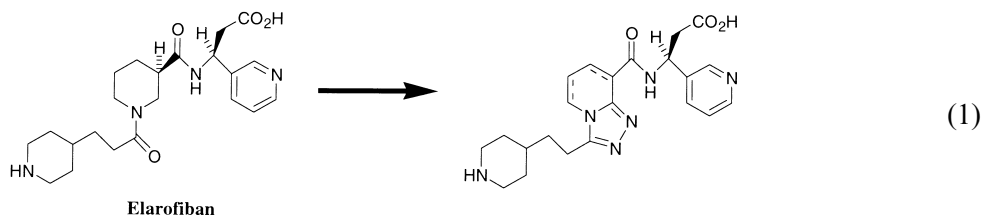
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Abstract

Amide derivatives of cyclic amines are common templates for biologically active chemical entities. To constrain the tertiary amide moiety of *N*-acylpiperidine-based GPIIb/IIIa antagonists for further structure–activity study, we have prepared new 1,2,4-triazolo[4,3-*a*]pyridines. The syntheses of two classes of triazolopyridines are reported. © 2000 Elsevier Science Ltd. All rights reserved.

Piperidine-based amides are important templates in biologically active chemical systems.^{1–4} Some of their prominent therapeutic applications have been in the areas of kappa-opioid analgesics,⁵ antiinflammatory agents,⁶ and antithrombotic agents.⁷ We identified a novel series of antithrombotic agents based on nipecotic acid as a peptidomimetic scaffold, from which elarofiban⁸ (RWJ-53308) was selected for clinical evaluation in humans (Eq. (1)).^{9–11} Thus, as a follow-up, we sought to incorporate an element of planar, geometric rigidification into the key tertiary amide group. Possible improvements in the bioactivity and, especially, the pharmacokinetics of these fibrinogen receptor (GPIIb/IIIa) antagonists were envisioned by the use of such ring-fused analogues. In this letter, we report the synthesis of bicyclic triazolopyridines bearing functionality on both rings to accommodate the structure–activity relationship.⁸ Besides providing a basis for novel fibrinogen receptor antagonists, our results suggest a general annulation strategy for the preparation of variably functionalized, fused triazole scaffolds suitable for chemical library preparation by high-throughput organic synthesis.

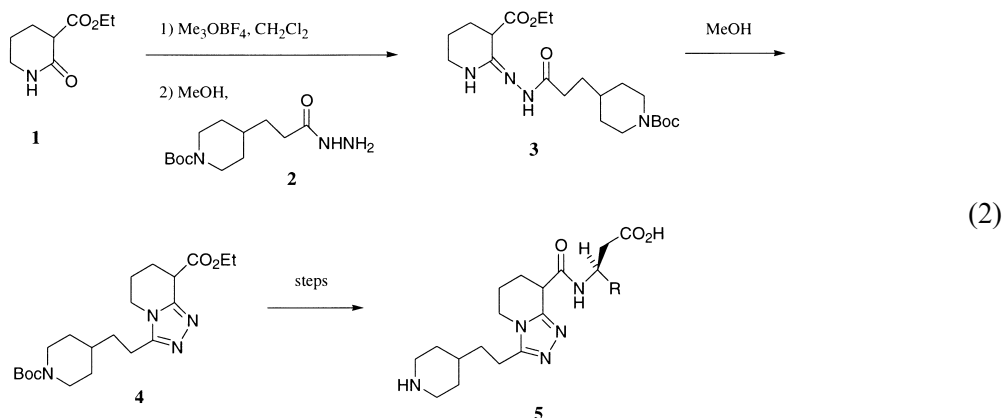


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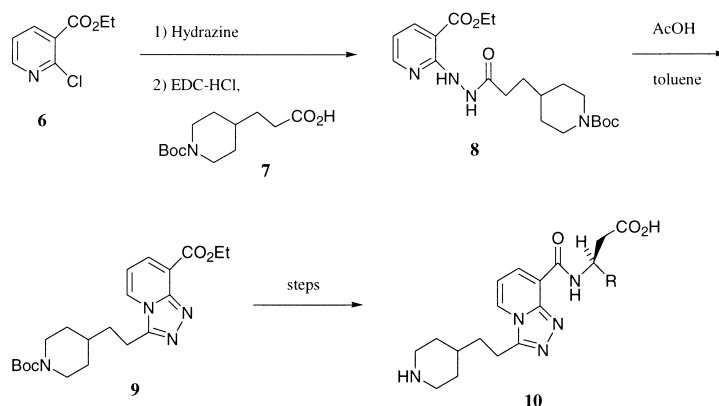
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Fusion of a five-membered ring at the N-1/C-2 position of nipecotic acid would elaborate a new bicyclic ring system and thereby remove a potentially metabolically labile tertiary amide (Eq. (1)). A synthetic conversion of 2-pyrrolidinones and acylhydrazides to substituted triazolopyrroles reported by Rigo et al. provided a precedent for our nipecotic acid-type system.^{12,13} In Rigo's three-step procedure, 5-carboxy-2-pyrrolidinone was *O*-methylated with dimethyl sulfate, treated with *N*-acetylhydrazide, and warmed to give an amidrazone intermediate, which was heated with chlorotrimethylsilane/triethylamine to afford the bicyclic triazole.

Disubstituted tetrahydrotriazolopyridines¹⁴ were prepared in a similar fashion. A reliable, high-yielding method utilizing existing 3-(4-piperidine)propionic acid and enantiomerically enriched β -amino acid intermediates (R = aryl) was desired to synthesize a number of analogues on a gram scale.¹⁵ To this end, commercially available piperidone **1** was methylated with trimethyloxonium tetrafluoroborate in dichloromethane and treated with Boc-protected acylhydrazide **2** in methanol to give an amidrazone intermediate **3** (Eq. (2)). Rigo's annulation method, which had required basic, chlorotrimethylsilane-mediated conditions at high temperatures, produced a complex mixture of products in our hands when starting with the five-membered ring system, ethyl 2-oxo-3-pyrrolidine carboxylate. By contrast, simply warming the corresponding six-membered ring amidrazone **3** in methanol under neutral conditions for an extended period of time (18–24 h) afforded tetrahydrotriazolopyridine **4** in good yield. The corresponding *N*-benzyloxycarbonyl-protected analogue of **4** was prepared in a similar fashion.



A triazolopyridine ring system with six-membered ring unsaturation¹⁶ was also prepared to introduce further conformational rigidification of the central scaffold. To address this modification, intermediate ester **9** was prepared in three steps starting with chloronicotinate **6** (Eq. (3)).¹⁷ Initial attempts to add hydrazide **2** directly to **6** (as in Eq. (2)) to provide acylhydrazide **8** proved unsuccessful. Alternatively, compound **6** was treated with hydrazine in warm dioxane followed by water soluble carbodiimide-mediated (EDC) formation of acylhydrazide **8**. Generation of triazolopyridine **9** by cyclization of acylhydrazide **8** utilizing neutral methanol conditions was unsuccessful in this case. Instead, the hydrazide was cyclized under acidic conditions (acetic acid/toluene, reflux) to afford key intermediate **9**.



In conclusion, we have described concise syntheses of two novel, disubstituted 1,2,4-triazolo[4,3-*a*]pyridines. Given the straightforward, three-step assembly from a key pyridyl-acylhydrazide intermediate, this methodology represents a general approach to new peptidomimetics¹⁸ of tertiary amides. As an application of this chemistry, esters **4** and **9** were carried forward to the biologically interesting antithrombotic compounds **5** and **10**.^{19,20}

Acknowledgements

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14. For a patent on 1,2,4-triazolo[4,3-*a*]pyridine-8-acetic acids as angiotensin II receptor antagonists, see: Manning, R. E.; Huang, H. C. US Patent 5,426,105, 1995.
15. Ester **4** was synthesized from **1** as follows: Trimethyloxonium tetrafluoroborate (5.46 g, 36.9 mmol) was added to a solution of ethyl 2-oxo-3-piperidine carboxylate (6.31 g, 36.9 mmol) in CH₂Cl₂ (62 mL) and the reaction was stirred for 18 h. *N*-Boc-3-(4-piperidine)propionic hydrazide (10.0 g, 36.9 mmol) in CH₂Cl₂ (62 mL) was added. The resulting mixture was stirred for 24 h, diluted with CH₂Cl₂ (150 mL), and washed with brine (2×50 mL). The organic layer was dried (Na₂SO₄) and evaporated to give a white solid, which was dissolved in MeOH (365 mL) and heated at reflux for 24 h. The mixture was cooled, concentrated in vacuo, and purified by flash chromatography (CH₂Cl₂:MeOH:NH₄OH, 97:2:1) to give **4** as a white solid (7.79 g, 52% yield, >98% purity). ¹H NMR (CDCl₃) δ 4.1 (q, 2H), 3.9 (m, 4H), 2.7 (m, 2H), 2.0 (m, 6H), 1.6 (m, 6H), 1.6 (s, 9H), 1.2 (t, 3H), 1.1 (m, 2H); ES/MS *m/z* 407.4 (MH⁺).
16. Jones, G.; Sliskovic, R. *Adv. Heterocycl. Chem.* **1983**, *34*, 79.
17. Ester **9** was synthesized from **6** as follows: Ethyl 2-chloronicotinate (2.1 g, 11.3 mmol) and anhydrous hydrazine (0.4 mL, 12.7 mmol) were heated in dioxane (54 mL) at 60°C for 2 h. The mixture was cooled to 23°C and concentrated in vacuo to give ethyl 2-hydrazinenicotinate (2.0 g, 97%). *N*-Boc-3-(4-piperidine)propionic acid (**7**; 2.85 g, 11.1 mmol) was dissolved in CH₂Cl₂ (55 mL), cooled to 0°C, and treated with ethyl 3-dimethylamino-propylcarbodiimide-HCl (2.55 g, 13.3 mmol), *N*-methylmorpholine (1.5 mL, 13.6 mmol), and 1-hydroxybenzotriazole (0.15 g, 1.11 mmol). Ethyl 2-hydrazinenicotinate (2.0 g, 11.0 mmol) was added and the reaction was briefly stirred at 0°C, warmed to 23°C, and stirred for 17 h. The mixture was diluted with CH₂Cl₂ (100 mL) and washed with water (3×50 mL). The organic layer was dried (MgSO₄) and concentrated in vacuo to give **8** as a white solid (2.24 g, 48%), which was heated with acetic acid (6.0 mL, 105 mmol) in toluene (106 mL) at reflux for 22 h using a Dean-Stark trap containing 4 Å molecular sieves. The mixture was concentrated in vacuo and flash chromatographed (CH₂Cl₂:MeOH, 98:2) to give **9** as a white powder (1.54 g, 77% yield, >98% purity). ¹H NMR (CDCl₃) δ 8.6 (d, 1H), 8.1 (d, 1H), 7.4 (dd, 1H), 4.6 (m, 2H), 4.4 (q, 2H), 4.1 (q, 2H), 3.5 (m, 1H), 3.2 (t, 2H), 2.7 (m, 2H), 1.8 (m, 4H), 1.6 (s, 9H), 1.3 (t, 3H); ES/MS *m/z* 403.4 (MH⁺).
18. Kelly, J. W. *Bioorg. Med. Chem.* **1999**, *7*, 1.
19. The biological results will be reported separately.
20. Hoekstra, W. J.; Lawson, E. C.; Maryanoff, B. E. WO 00/06570.